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Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.043
 wR factor = 0.123
Data-to-parameter ratio = 11.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

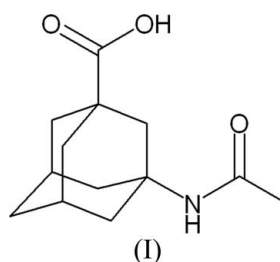
3-Acetamidoadamantane-1-carboxylic acid

The title compound, $\text{C}_{13}\text{H}_{19}\text{NO}_3$, crystallizes with two molecules, *A* and *B*, in the asymmetric unit. Intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds link symmetry-related molecules into infinite chains parallel to the $[101]$ direction, with short $\text{O}\cdots\text{O}$ distances of 2.534 (2) and 2.618 (2) Å in the chains of *A* and *B* molecules, respectively. Intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds cross-link these two chains.

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Comment

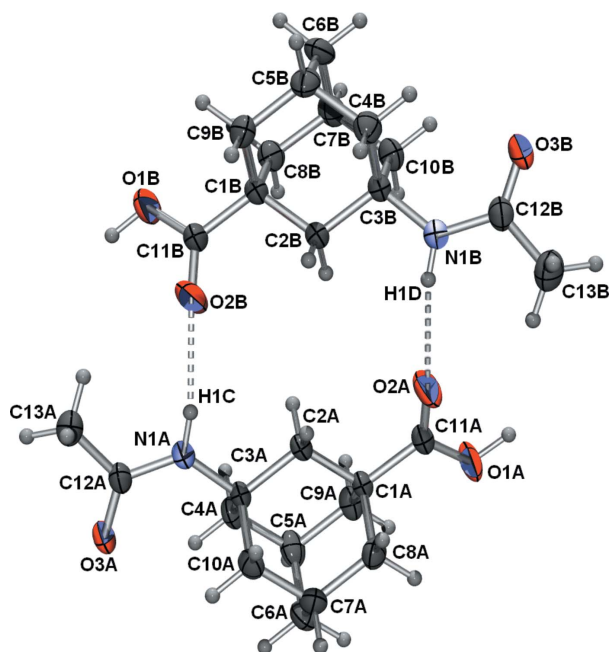
Acetamido derivatives of adamantane have been studied intensively, mainly due to their biological activity (Geigy, 1966). The title compound, (I) (Fig. 1), is a useful intermediate in the preparation of unnatural amino acids, such as 3-amino-1-adamantanecarboxylic acid. This acid has been incorporated into small peptides which showed antitumor activity *in vitro* (Horvat *et al.*, 2006). Here we report the crystal structure of (I).



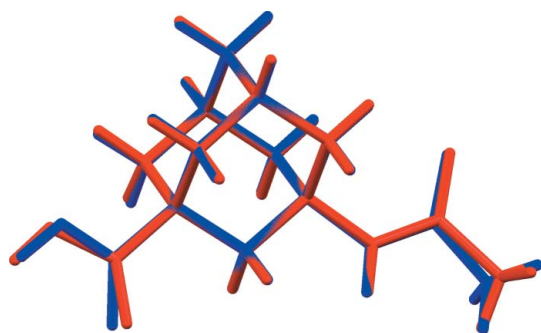
The title compound crystallizes in the space group $P2_1/n$, with two independent molecules, *A* and *B*, in the asymmetric unit. Most geometric parameters of *A* and *B* coincide within three standard uncertainties (Fig. 2) and agree with the literature data for the same atom types and hybridizations (Allen *et al.*, 1987). The only significant differences ($\delta \geq 3\sigma$) are those involving the torsion angles defining the orientations of the acetamide and carboxyl units (Table 1).

In the early stages of data reduction and structure solution, *E*-statistics revealed a hypercentric structure. Two molecules, *A* and *B*, which are related by a pseudo-inversion centre at $ca(1/5, 1/4, 1/4)$ (Fig. 3) generate hypersymmetry. When transformed into a standard setting ($P2_1/c$), the spatial arrangement of the molecules in the unit cell (Fig. 3) and the location of the pseudo-inversion centre agree with those observed for $P2_1/c$ structures with $Z' = 2$ (Zorky, 1996; Pidcock, 2006).

The crystal packing is dominated by double hydrogen-bonded chains running in the $[101]$ direction (Fig. 3) generated by pseudo-inversion centres. There are four symmetry-independent hydrogen bonds (Table 2), grouped into two pairs related by a pseudo-inversion centre. Two $\text{O}-\text{H}\cdots\text{O}$ bonds

**Figure 1**

The asymmetric unit of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability and H atoms are depicted as spheres of arbitrary radii. Intermolecular hydrogen bonds between molecules *A* and *B*, related by a pseudo-inversion centre, are shown as dashed lines.

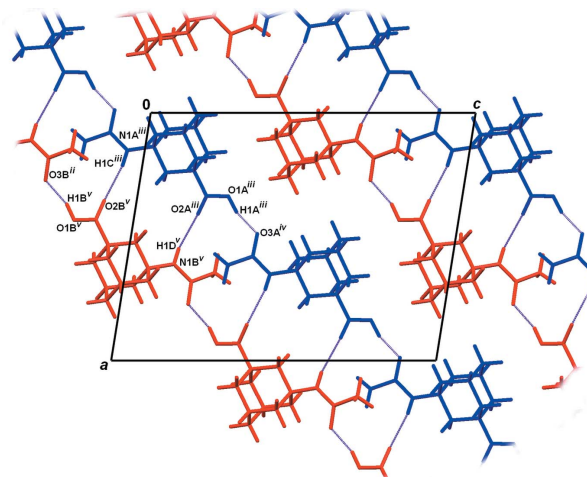
**Figure 2**

Overlay of molecules *A* (blue) and *B* (red); small conformational differences involve acetamido and carboxyl groups (Table 1).

are relatively short for their type and can be classified as medium-strong to strong according to Jeffrey (1997). The hydrogen bonds between the hydroxyl and amide sp^2 O atoms (O—H...O) link symmetry-equivalent molecules ($A \cdots A$; $B \cdots B$) while pairs of amide and carboxyl sp^2 O atoms (N—H...O) cross-link $A \cdots B$ molecules. The hydrogen-bonded units create a hydrophilic core, whereas the adamantyl units are located at the outer parts of the hydrogen-bonded chains. Alternating hydrophilic and hydrophobic regions characterize the molecular assembly in the solid state (Fig. 3).

Experimental

The title compound, (I), was prepared by treating adamantane-1-carboxylic acid with oleum in acetonitrile (Novikov *et al.*, 1980). After

**Figure 3**

Crystal packing of (I) in the *ac* plane with hydrophilic interiors of hydrogen-bonded chains and hydrophobic adamantyl groups. Molecules *A* are blue, *B* are red, and hydrogen bonds are violet. [Symmetry codes: (ii) $x - \frac{1}{2}, \frac{3}{2} - y, z - \frac{1}{2}$; (iii) $x - \frac{1}{2}, \frac{3}{2} - y, \frac{1}{2} + z$; (iv) $x, y, 1 + z$; (v) x, y, z .]

work-up, single crystals were obtained from the crude product by slow evaporation of a methanol solution.

Crystal data

$C_{13}H_{19}NO_3$
 $M_r = 237.29$
 Monoclinic, $P2_1/n$
 $a = 13.3487$ (6) Å
 $b = 10.5295$ (5) Å
 $c = 17.2793$ (7) Å
 $\beta = 99.073$ (3)°
 $V = 2398.31$ (18) Å³

$Z = 8$
 $D_x = 1.314$ Mg m⁻³
 Cu $K\alpha$ radiation
 $\mu = 0.76$ mm⁻¹
 $T = 293$ (2) K
 Prism, colourless
 $0.3 \times 0.15 \times 0.15$ mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: ψ scan
 (8 reflections; North *et al.*, 1968)
 $T_{min} = 0.845$, $T_{max} = 0.892$
 5262 measured reflections

5043 independent reflections
 3856 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.014$
 $\theta_{max} = 76.5^\circ$
 3 standard reflections
 frequency: 120 min
 intensity decay: 2%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.123$
 $S = 1.03$
 5043 reflections
 460 parameters
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0561P)^2 + 0.8403P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.28$ e Å⁻³
 $\Delta\rho_{min} = -0.30$ e Å⁻³
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.00042 (10)

Table 1

Selected torsion angles (°).

C13A—C12A—N1A—C3A	−174.35 (16)	C13B—C12B—N1B—C3B	171.35 (16)
O3A—C12A—N1A—C3A	3.9 (3)	O3B—C12B—N1B—C3B	−6.2 (3)
C2A—C1A—C11A—O1A	164.46 (15)	C2B—C1B—C11B—O1B	−168.34 (15)
C2A—C1A—C11A—O2A	−17.9 (2)	C2B—C1B—C11B—O2B	12.7 (2)

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1A-H1A\cdots O3A^i$	0.95 (3)	1.61 (3)	2.534 (2)	163 (3)
$O1B-H1B\cdots O3B^{ii}$	0.92 (4)	1.73 (3)	2.618 (2)	163 (3)
$N1A-H1C\cdots O2B^{iii}$	0.85 (2)	2.17 (2)	2.976 (2)	158 (2)
$N1B-H1D\cdots O2A^{iv}$	0.85 (2)	2.22 (2)	3.006 (2)	154 (2)

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$; (ii) $x - \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$; (iii) $x + \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$; (iv) $x - \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$.

H atoms were located in a difference Fourier map and refined freely.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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